

# Structural and electrical properties of poly(3-octylthiophene) (P3OT) films: A scanning probe microscopy study.

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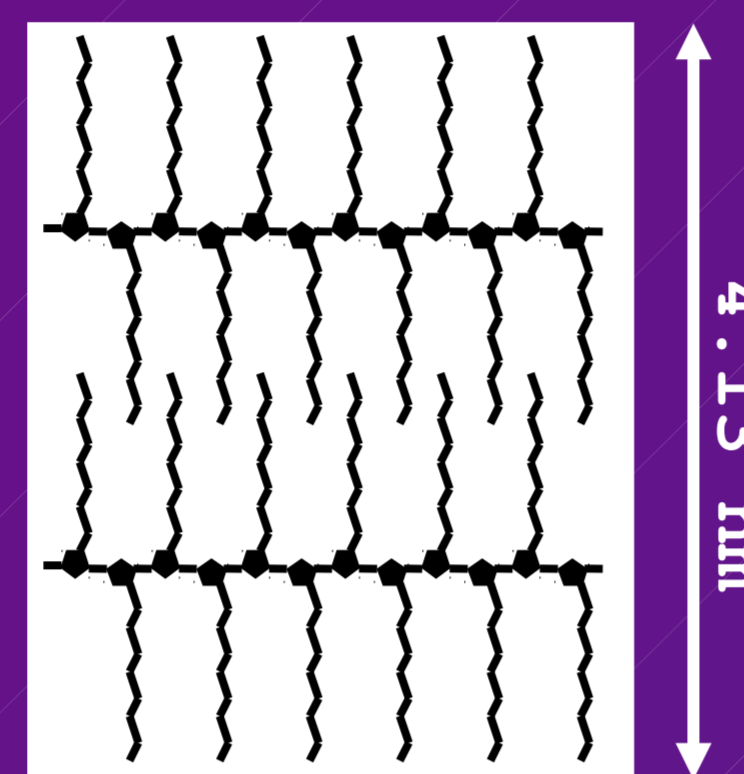
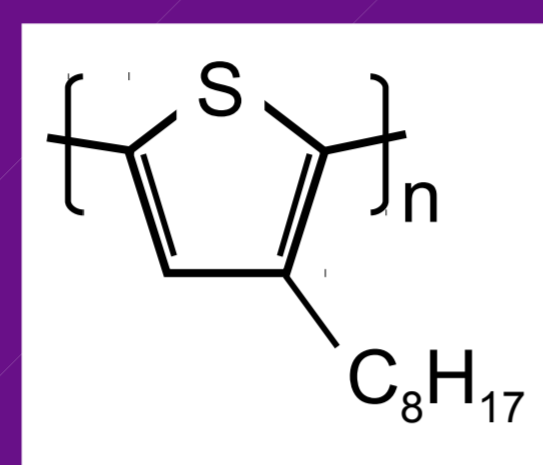
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## Introduction

Poly(3-octylthiophene) is one of the most promising materials for applications in organic opto-electronic devices such as field-effect transistors, light-emitting diodes or plastic solar cells due to its high hole mobility. Understanding the correlation between the morphological and electrical properties is of vital importance, since the structural arrangements of the molecules determines and limits the overall devices. In the present work we use scanning force microscopy techniques to study P3OT thin films, combining topographic, electrostatic, capacitance and electronic transport measurements.

### Poly(3-octylthiophene)

### P3OT



Greater than 98.5% head to tail regioregular

Molecular weight Average  $M_w \sim 142000$

Average  $M_n \sim 54000$

Polydispersity index  $\sim 2.6$

Conductivity  $1 \times 10^{-6}$  S/cm

Our samples conductivity  $1 \times 10^{-5}$  S/cm

Solution of P3OT in Toluene

$\sim 5; 10; 20; 40$  mg/ml

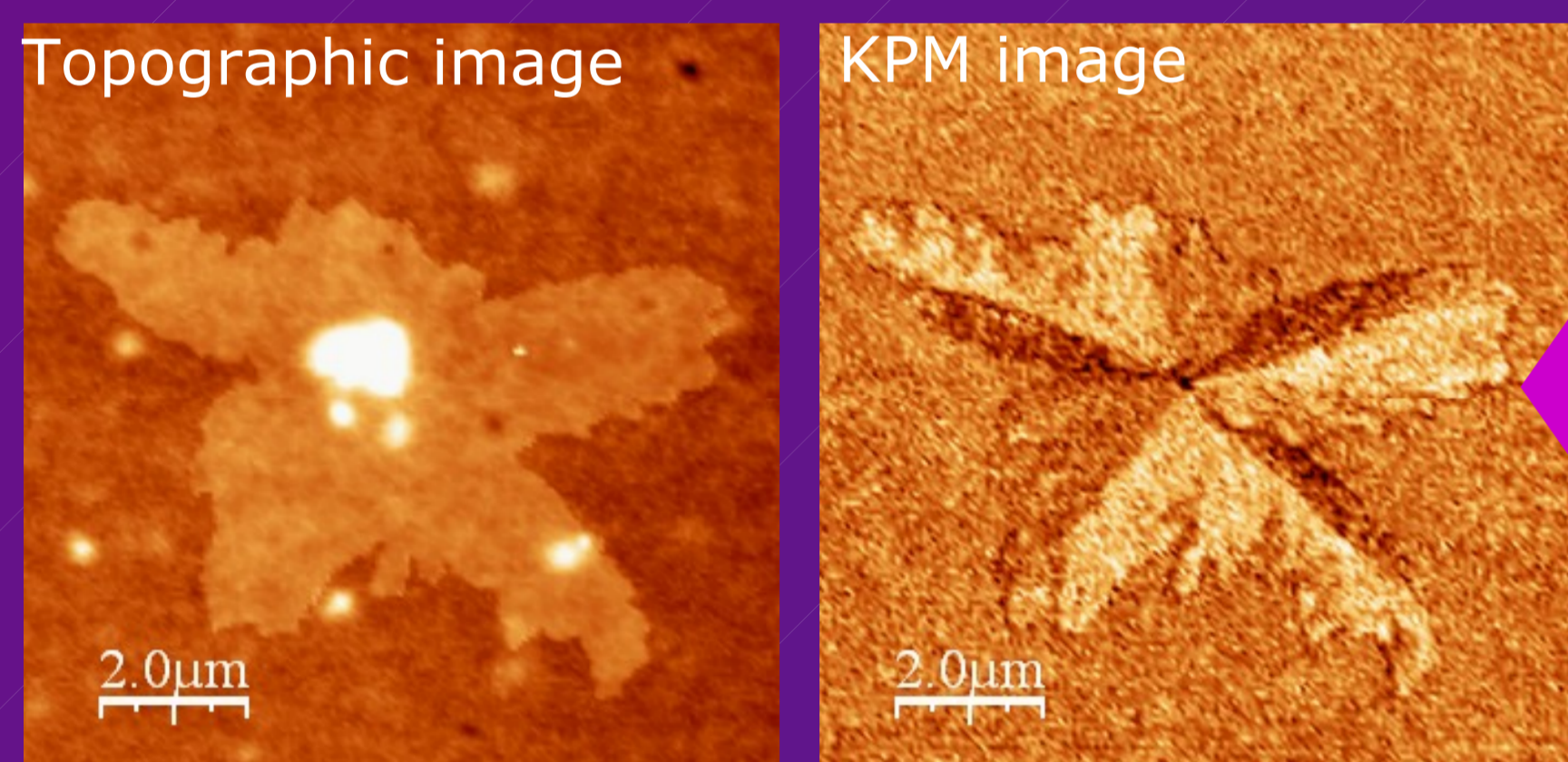
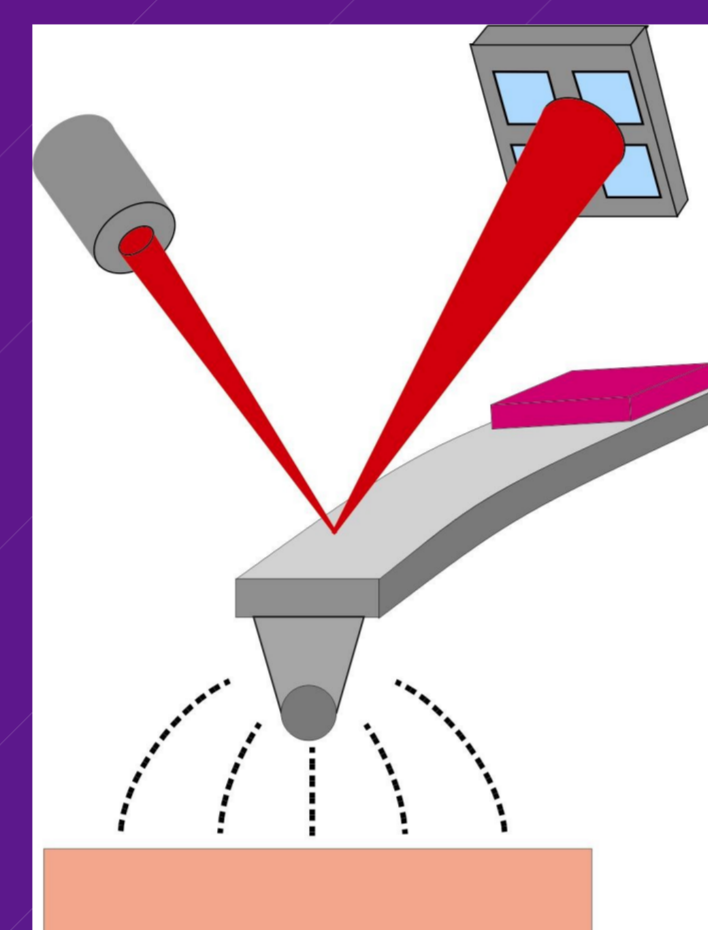
## Kelvin, Electrostatic and Capacitance

$$\text{Electrostatic interaction } W_{\text{electrostatic}} = \frac{1}{2} C(d) (V_{\text{bias}} - V_{\text{CP}})^2$$

$$W_{\text{electrostatic}}^{(1)} = F_n = \frac{1}{2} C'(d) (V_{\text{bias}} - V_{\text{CP}})^2 \quad W_{\text{electrostatic}}^{(2)} = \frac{1}{2} C''(d) (V_{\text{bias}} - V_{\text{CP}})^2$$

$$\Delta\omega = -\frac{\omega_{\text{res}}}{c_{\text{lever}}} \frac{1}{2} C''(d) (V_{\text{bias}} - V_{\text{CP}})^2 \quad V_{\text{bias}} = V_{\text{dc}} + V_{\text{ac}} \sin(\omega t)$$

$$\Delta\omega_{\text{res}}(\omega) \propto V_{\text{ac}} C''(d) (V_{\text{dc}} - V_{\text{CP}}) \quad \Delta\omega_{\text{res}}(2\omega) \propto V_{\text{ac}}^2 C''(d)$$

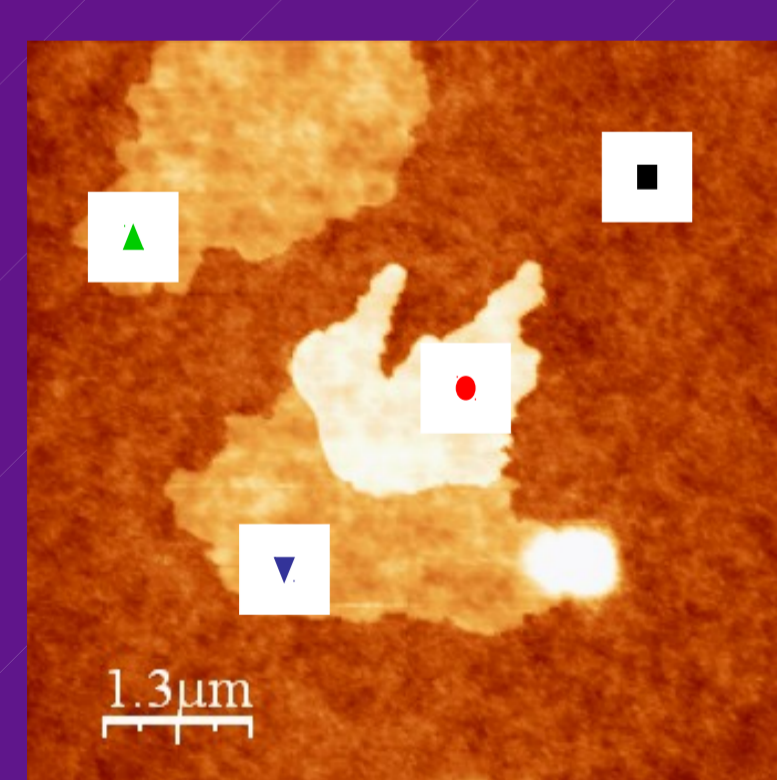


Different surface contact potential domains are shown

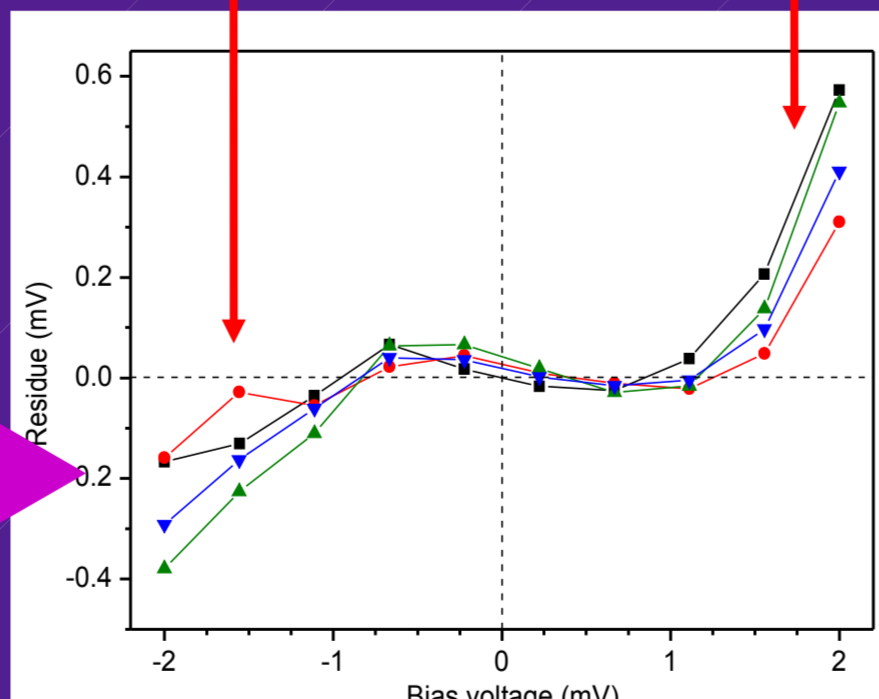
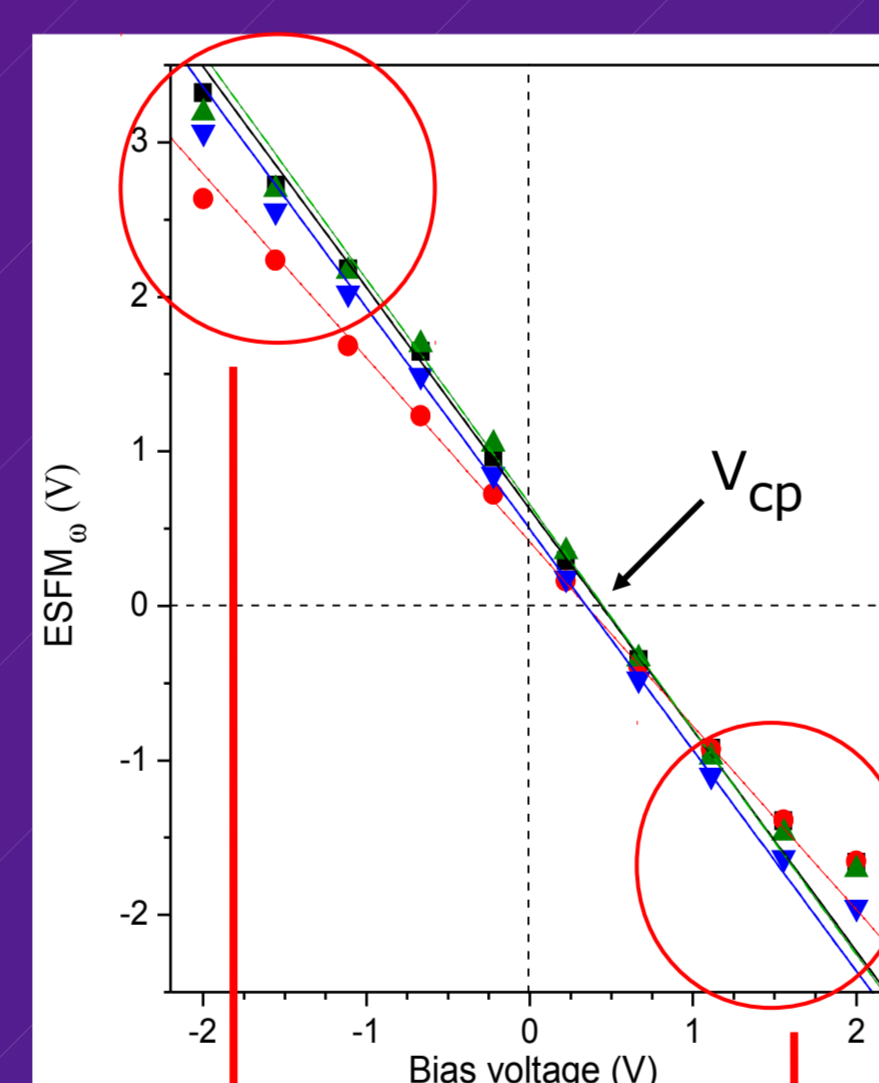
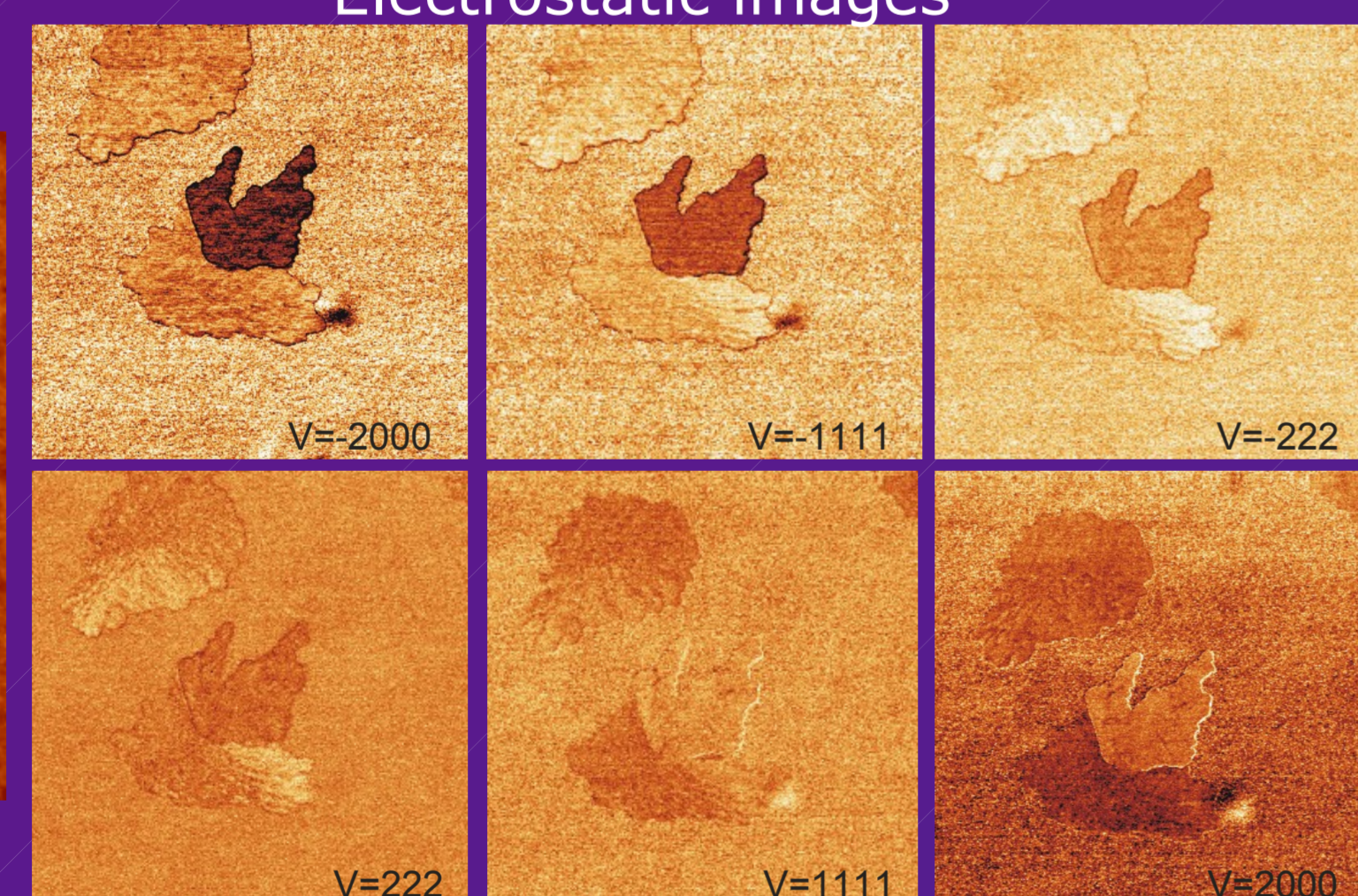
Zone	$\Delta\text{CP}(\omega)$ [mV]	$\Delta\text{CP}(\text{KPM})$ [mV]
■	0	0
●	93	32
▲	-10	-21
▼	43	25

20 mg/ml on gold thin film

Topographic image

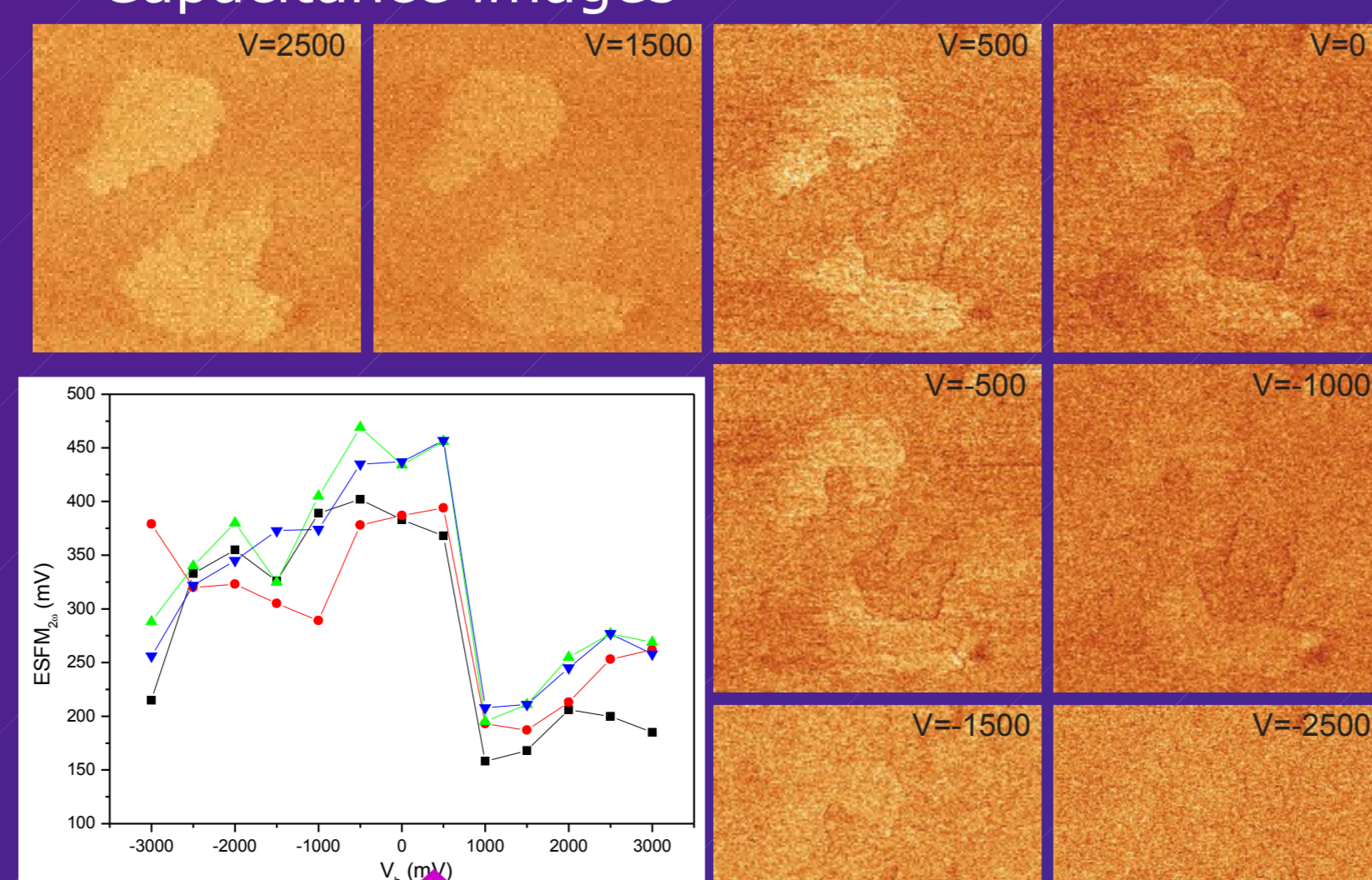


Electrostatic images



Deviation of the electrostatic model from the linearity with the applied voltage

Capacitance images



Capacitance varies with the applied voltage

## Conclusions

In this work we have carried out the morphological and electrical characterization of P3OT thin films.

Finding self-organized structures, which show different surface contact potential domains. There is a deviation of the electrostatic model from the linearity and a variation of the capacitance with the applied voltage. The last one probably due to band bending effects.

No appreciable resistance differences between the different polymer regions has been found.

UV irradiation induces holes and the growth of dendritic structures. The degradation is faster in the polymer background than in the layered structures.

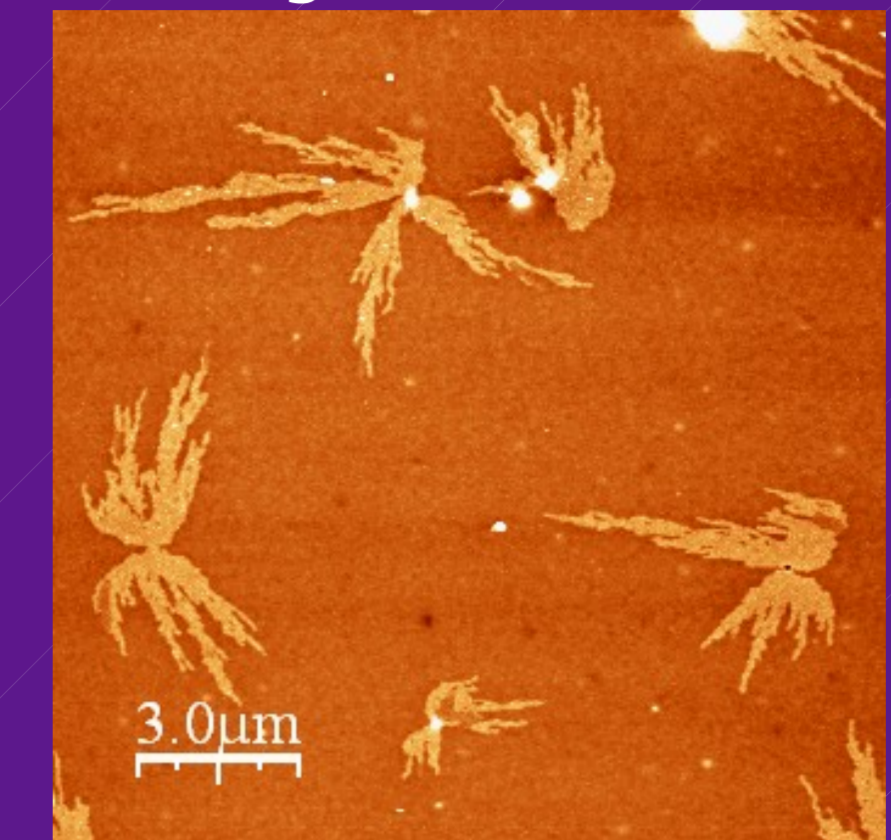
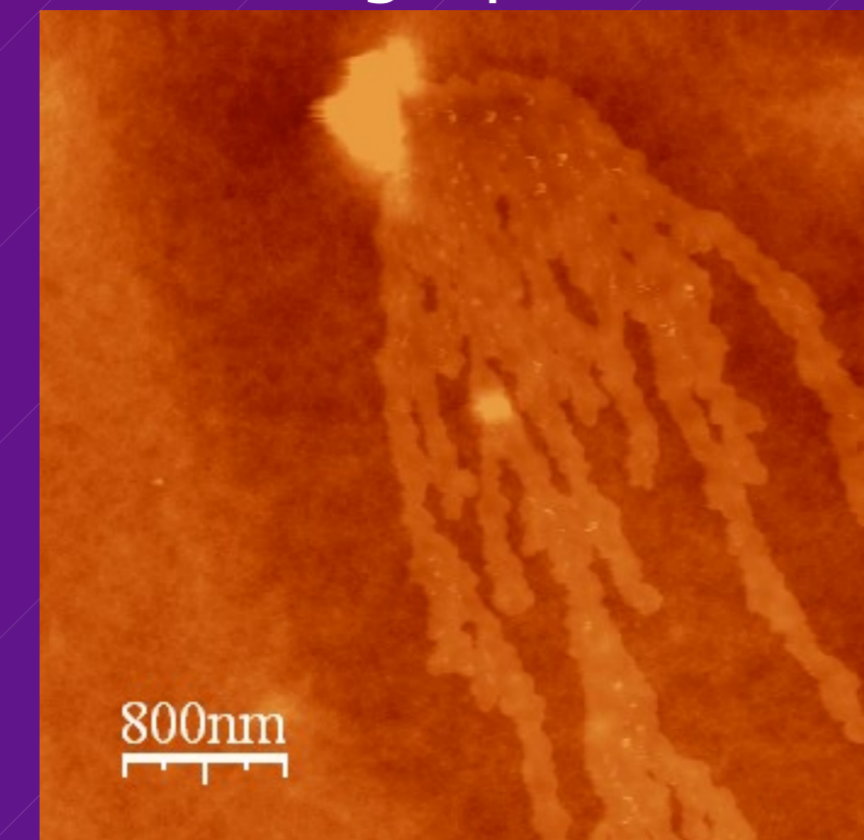
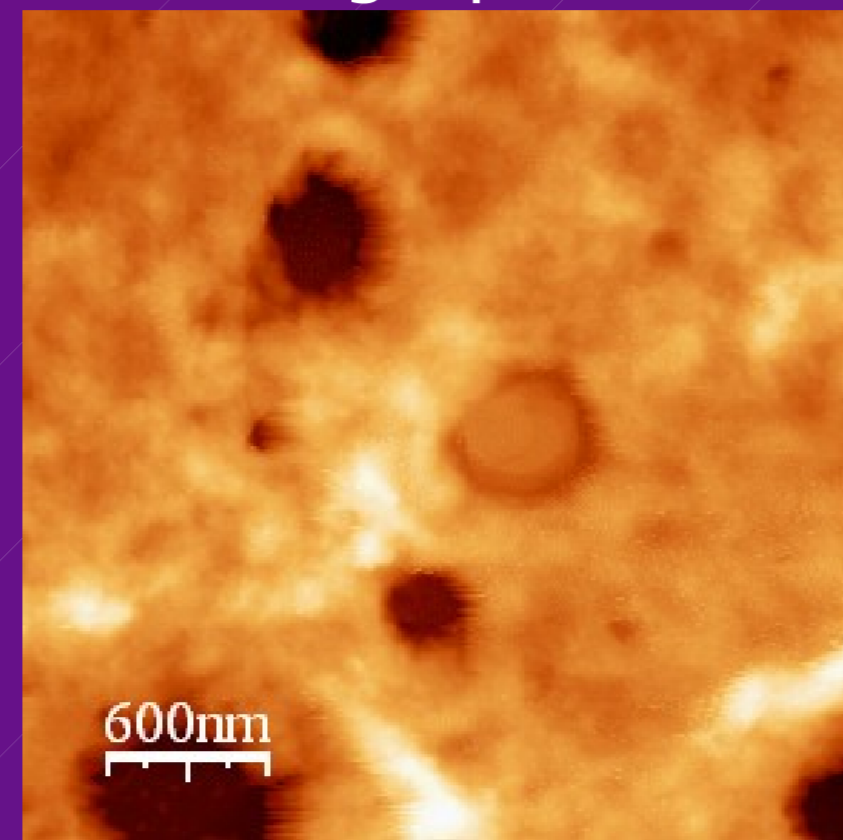
## Morphology

### Drop casting versus Spin coating

5 mg/ml Drop casted on graphite

20 mg/ml Spin coated on graphite

20 mg/ml Spin coated on gold thin film

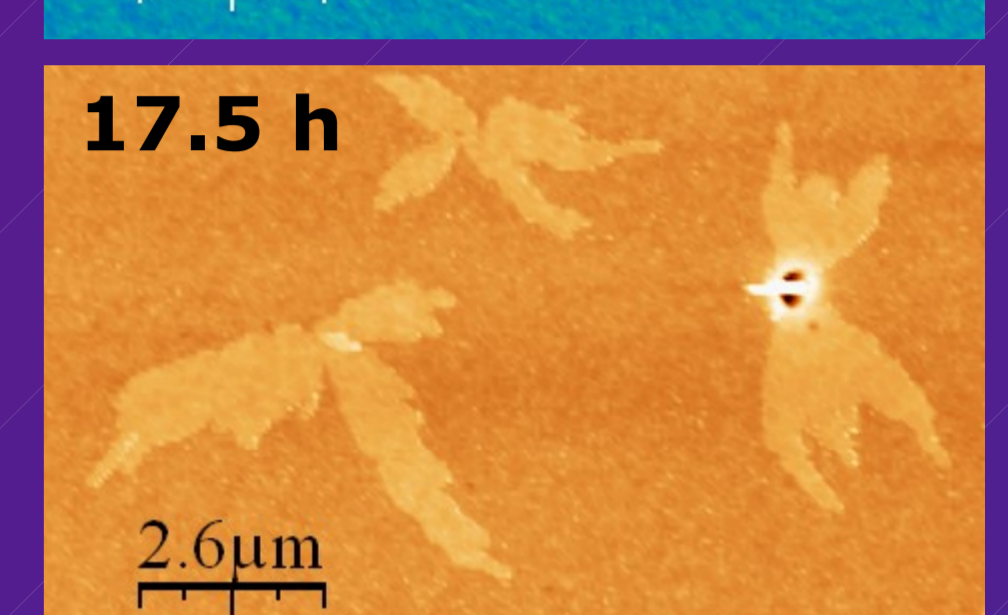
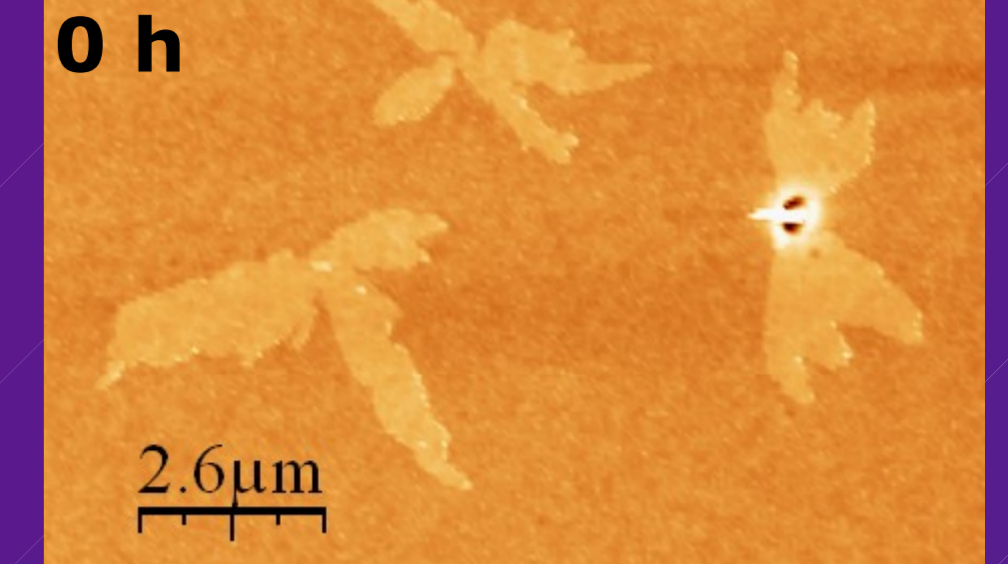


Rough morphologies with layered structures and many voids

Self-organized layered structures "jellyfishes" 1-4 layers of height 4-5 nm each one onto a disordered polymer layer

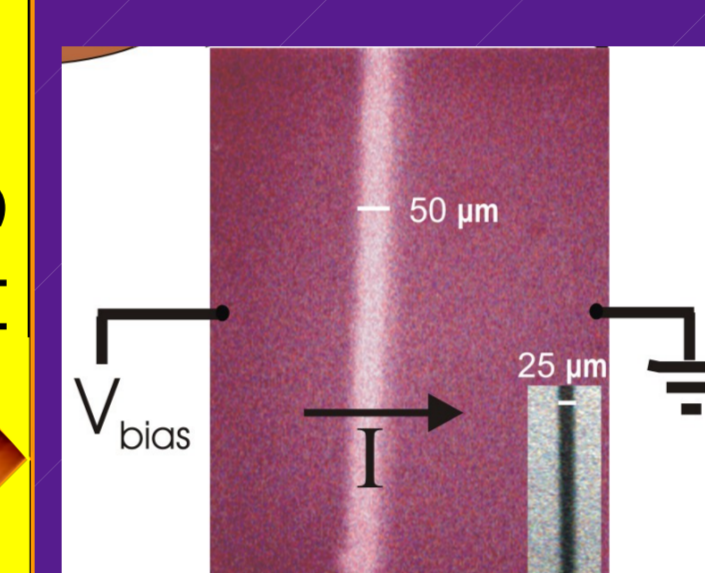
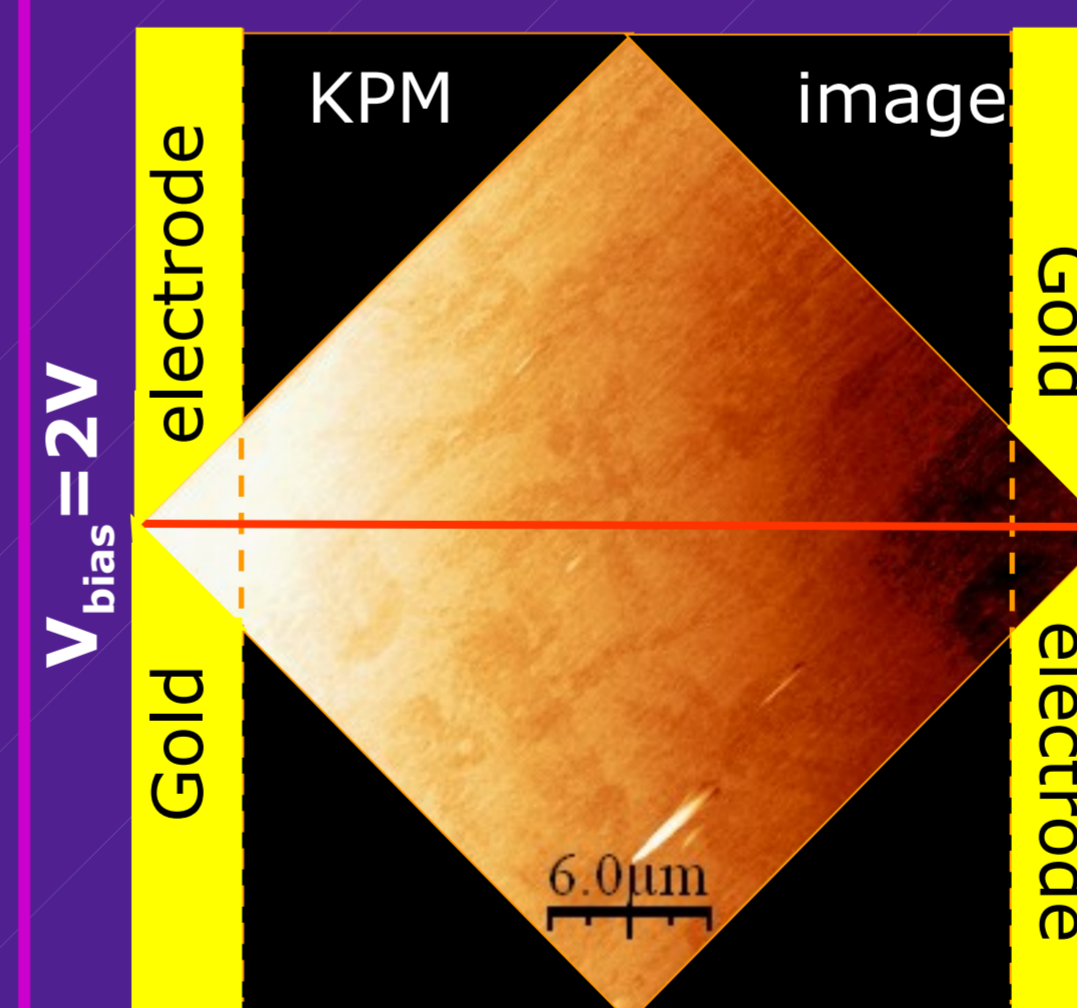
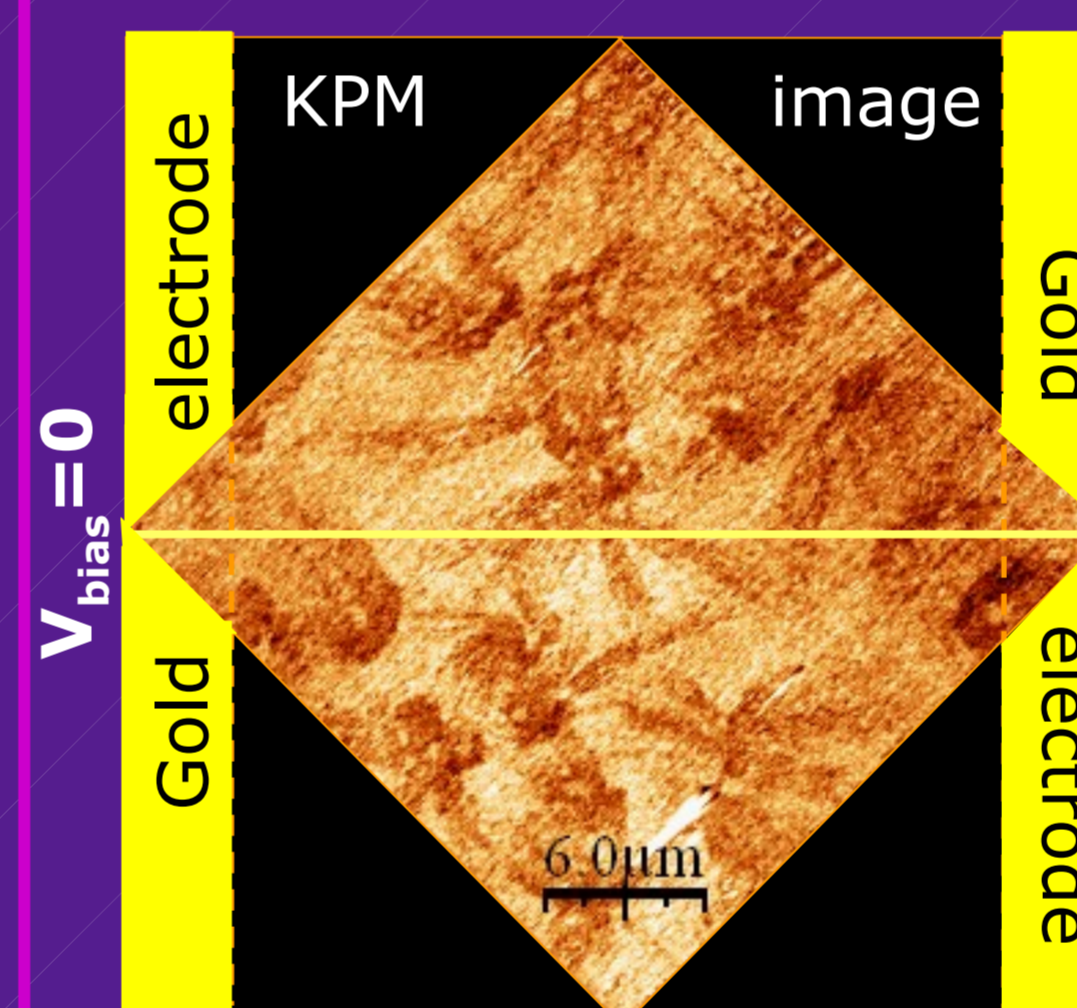
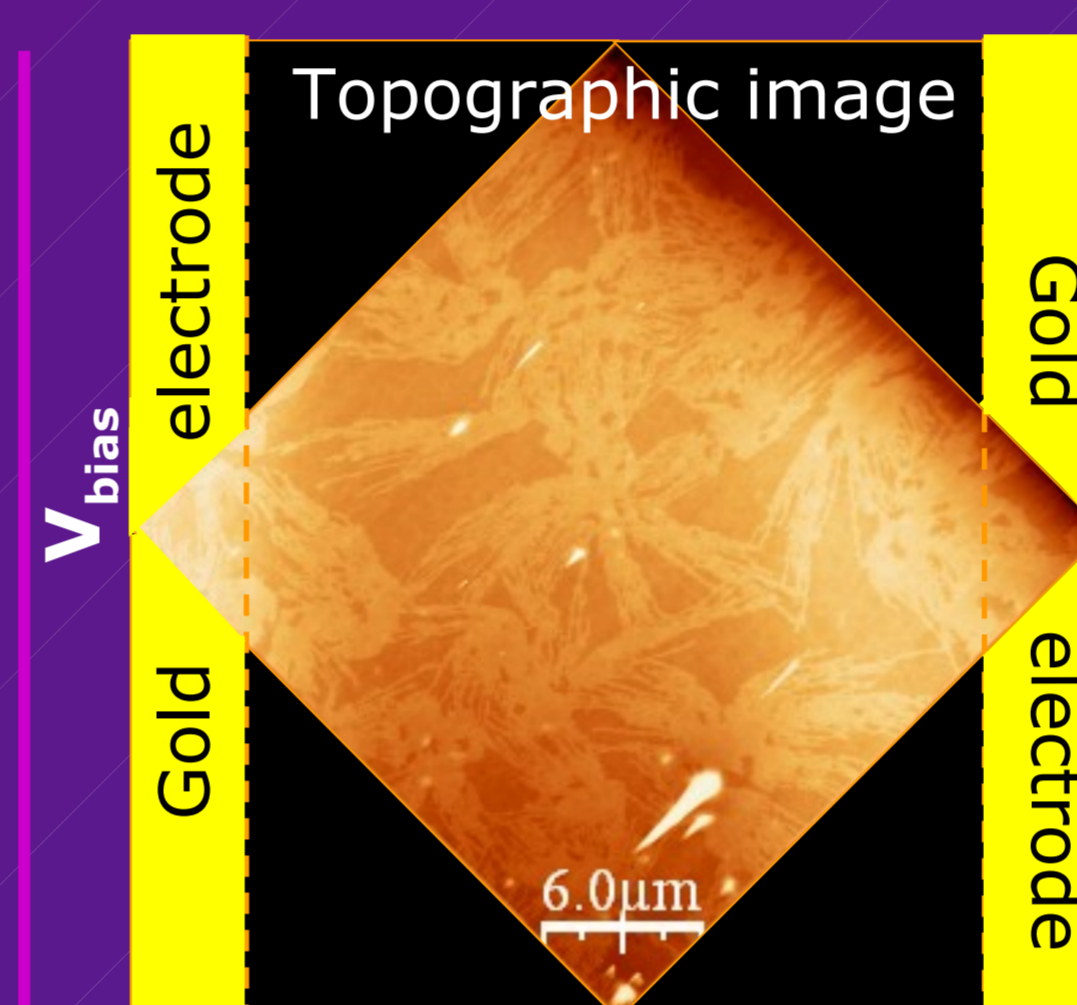
## Growth dynamic

20 mg/ml on glass cover



$V \sim 1-5 \mu\text{m}^2/\text{day}$

## KPM and transport measurements

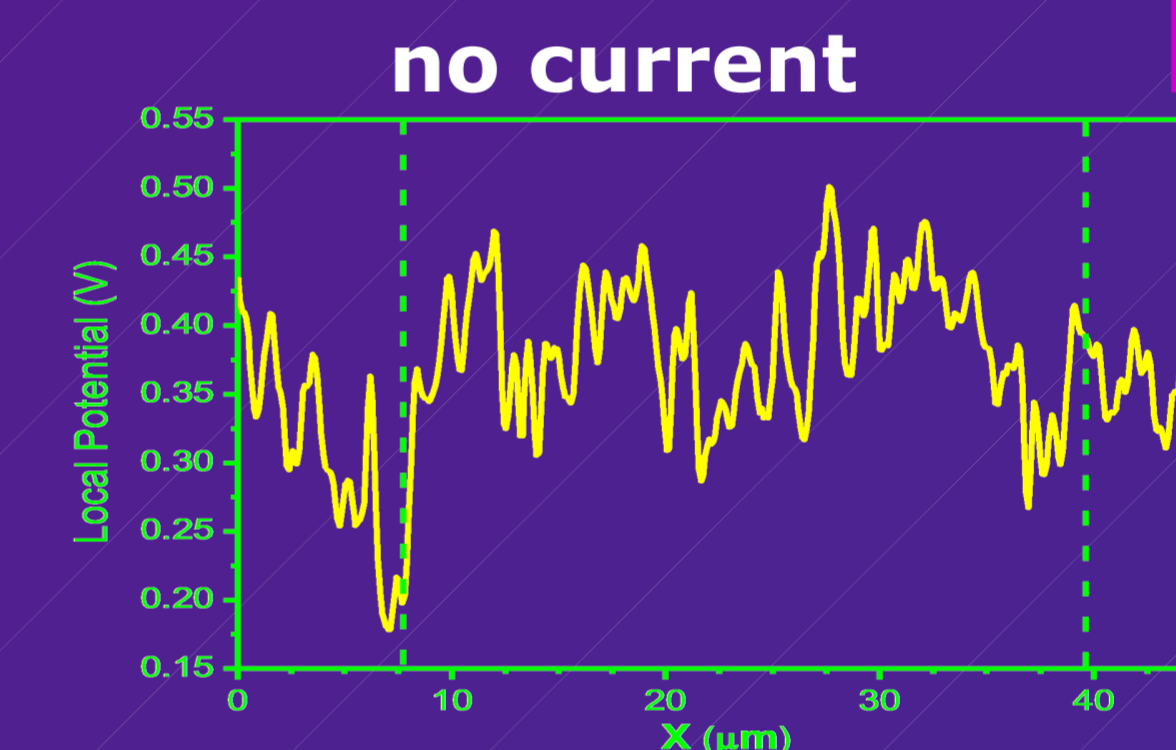


20 mg/ml

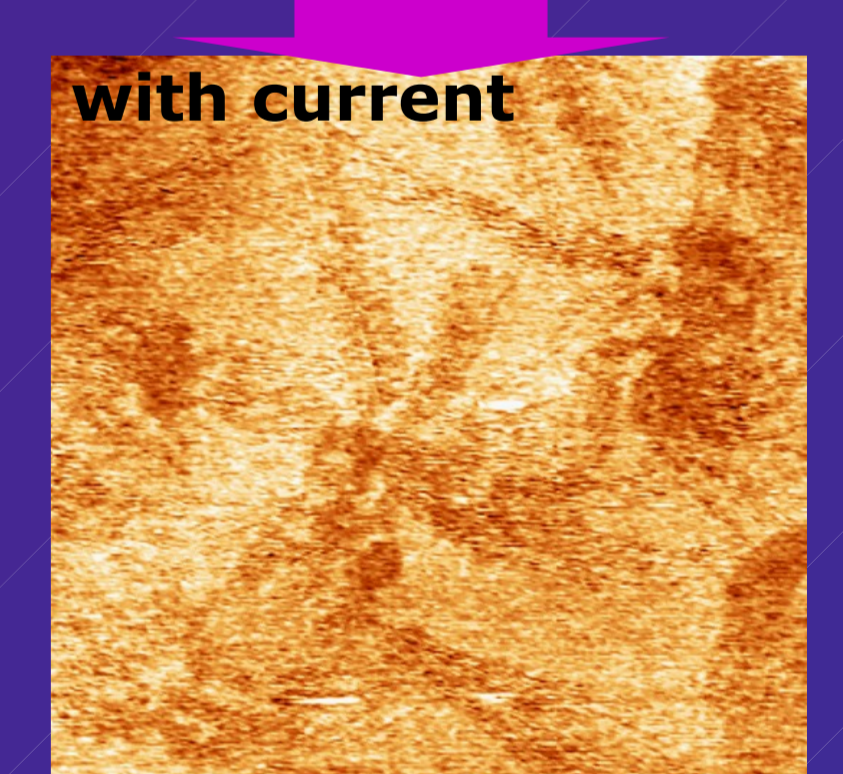
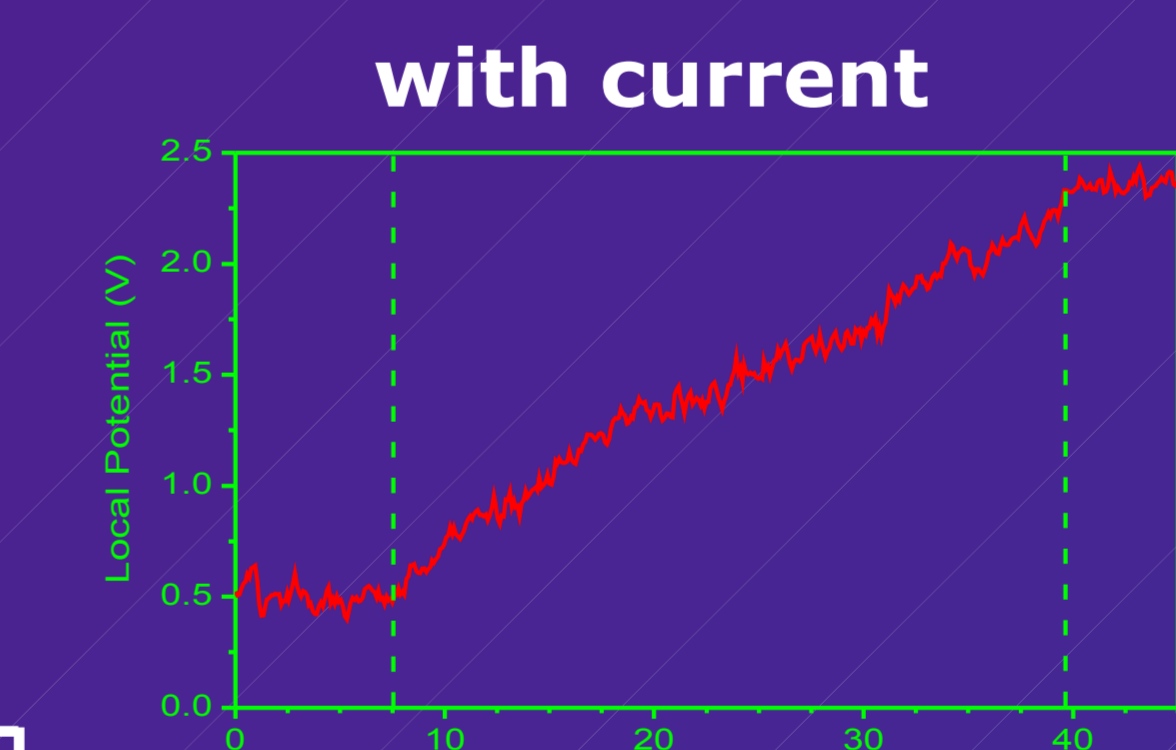
What we are measuring?

$V_{\text{bias}} = 0; I = 0$  Local Potential  $= V_{\text{CP}}$   
 $V_{\text{bias}} \neq 0; I \neq 0$  Local Potential  $= V_{\text{CP}} + I_{\text{loc}} R_{\text{loc}}$

After subtracting the contribution of the global mean potential



No differences due to resistance are found

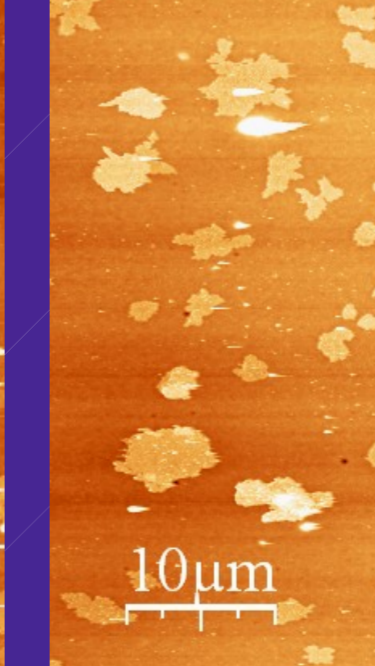


## Ultra-violet degradation

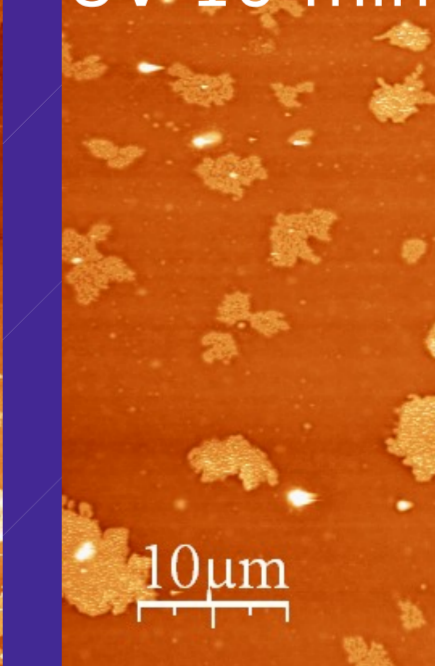
UV 1 min



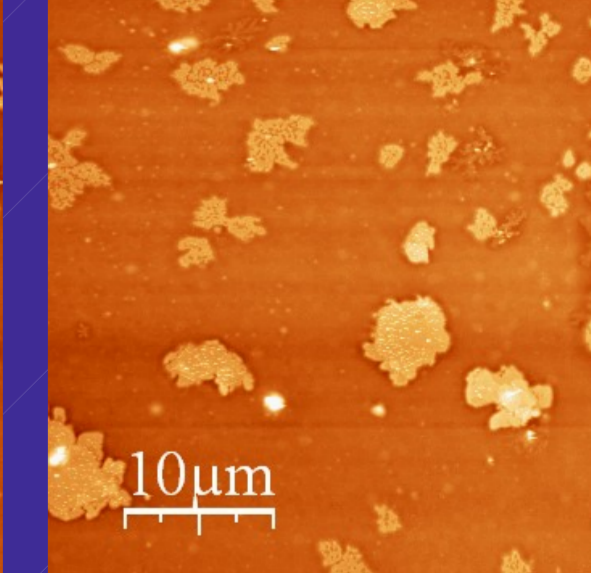
UV 5 min



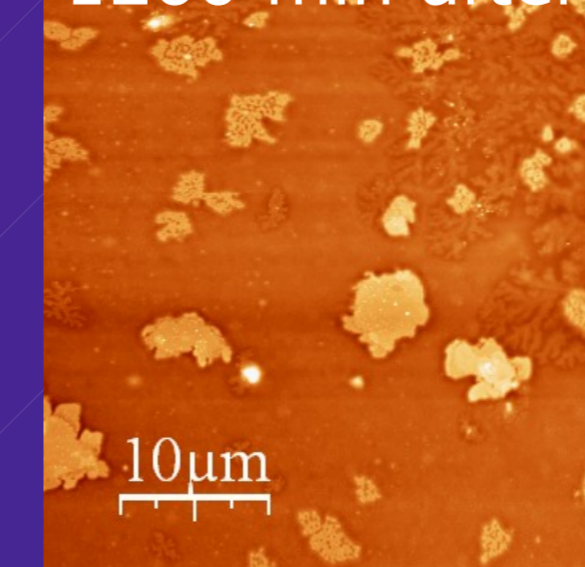
UV 10 min



UV 10 min 280 min after



UV 10 min 1260 min after

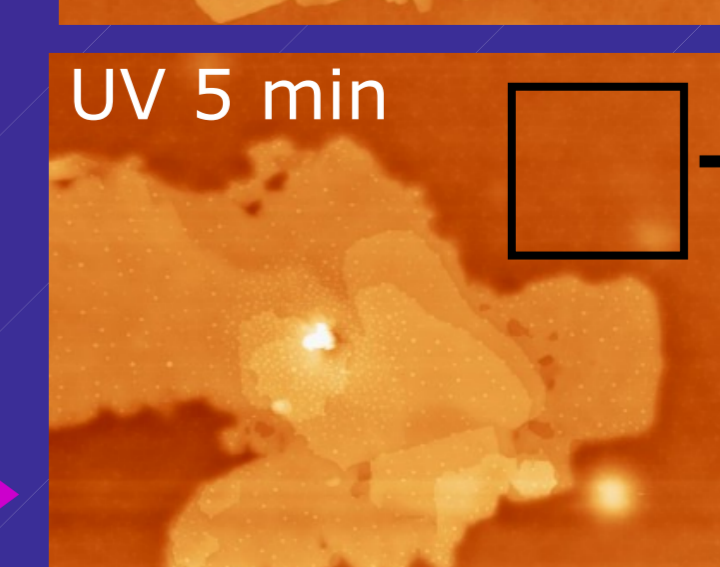
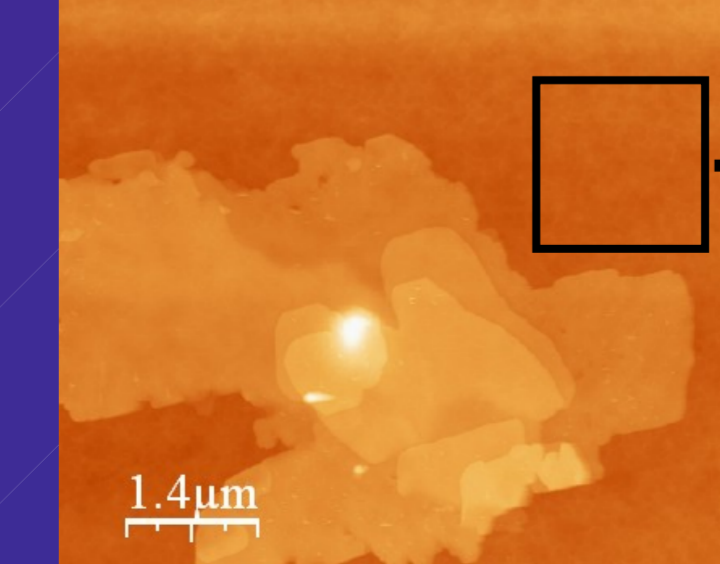


40 mg/ml on glass cover

Hole formation and growth of dendritic structures after the irradiation

$V \sim 630 \mu\text{m}^2/\text{day}$

Without UV



Faster degradation of the polymer background than the layered structures