

DEPARTAMENTO DE QUÍMICA INORGÁNICA

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<https://chem.libretexts.org/>

José Ruiz

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La dirección de acceso a la sección **Bookshelves** es <https://chem.libretexts.org/Bookshelves>.

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Bookshelves

Last updated: Mar 9, 2021

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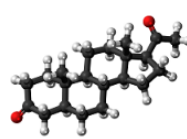
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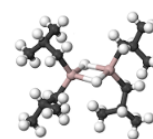
Introductory,
Conceptual, and GOB
Chemistry



General Chemistry



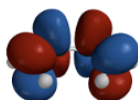
Organic Chemistry



Inorganic Chemistry



Analytical Chemistry



Physical & Theoretical
Chemistry




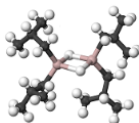
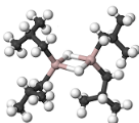
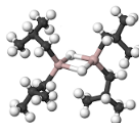
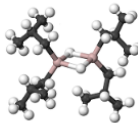
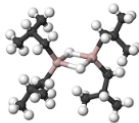
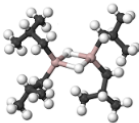
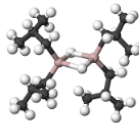


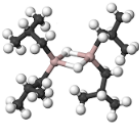
Biological Chemistry



Environmental
Chemistry

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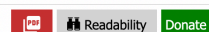
Como ejemplo, a la Materia de Química Inorgánica se puede acceder a través de https://chem.libretexts.org/Bookshelves/Inorganic_Chemistry

 Supplemental Modules and Websites (Inorganic Chemistry)	 Book: Bioinorganic Chemistry (Bertini et al.)	 Book: Introduction to Inorganic Chemistry (Wikibook)	 Book: Inorganic Chemistry (Saito)
 Book: Introduction to Organometallic Chemistry (Ghosh and Balakrishna)	 Chemistry of the Main Group Elements (Barron)	 Map: Inorganic Chemistry (Housecroft)	 Map: Inorganic Chemistry (LibreTexts) 
 Inorganic Coordination Chemistry (Landskron)		 Book: Principles of Inorganic Chemistry II (Nocera)	



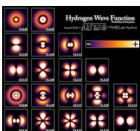
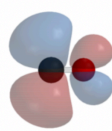
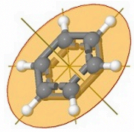
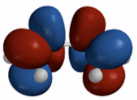
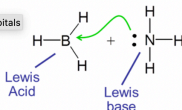
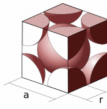
Además de libros, dispone también de mapas conceptuales (Maps), que pretenden reflejar cómo están relacionados las ideas y los conceptos. Un ejemplo de un "map" en construcción es el de Inorganic Chemistry (LibreTexts).

Map: Inorganic Chemistry (LibreTexts)

Last updated: May 1, 2022



This text follows the organization around Miessler, Fischer and Tarr's textbook, but with content built as an Open Education Resource. This text map is under construction. If you'd like to organize with us to pitch in, check out the under construction [spreadsheet here](#) (click) or email Kathryn Haas (khaas@saintmarys.edu).

 Front Matter	 1: Introduction to Inorganic Chemistry	 2: Atomic Structure	 3: Simple Bonding Theory
 4: Symmetry and Group Theory	 5: Molecular Orbitals	 6: Acid-Base and Donor-Acceptor Chemistry	 7: The Crystalline Solid State

Otros mapas están consolidados, como el Inorganic Chemistry (Housecroft): [https://chem.libretexts.org/Bookshelves/Inorganic_Chemistry/Map%3A_Inorganic_Chemistry_\(Housecroft\)](https://chem.libretexts.org/Bookshelves/Inorganic_Chemistry/Map%3A_Inorganic_Chemistry_(Housecroft)).

CHEMISTRY
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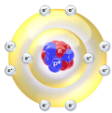


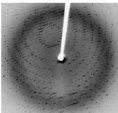
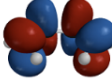


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Contents Home Bookshelves Inorganic Chemistry Map: Inorganic Chemistry (Housecroft)

Map: Inorganic Chemistry (Housecroft)

Last updated: Feb 3, 2021 Downloads Readability Donate

Inorganic chemistry deals with the synthesis and behavior of inorganic and organometallic compounds. This field covers all chemical compounds except the myriad organic compounds (carbon based compounds, usually containing C-H bonds), which are the subjects of organic chemistry.

 Front Matter	 1: Basic Concepts: Atoms	 2: Basic Concepts: Molecules	 3: Introduction to Molecular Symmetry
 4: Experimental Techniques	 5: Bonding in Polyatomic Molecules	 6: Structures and Energetics of Metallic and Ionic Solids	 7: Acids, bases and ions in aqueous solution




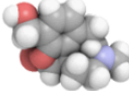


El Proyecto dispone también de acceso a herramientas didácticas (**Learning objects**) via https://chem.libretexts.org/Ancillary_Materials.

Contents Home Learning Objects

Learning Objects


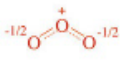

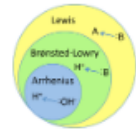
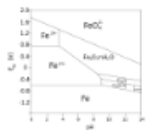
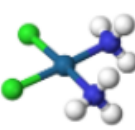
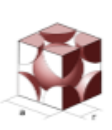
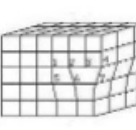
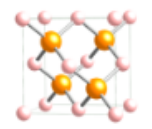
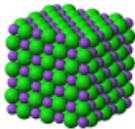
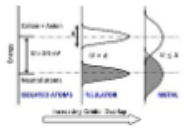
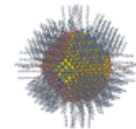
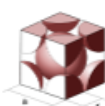
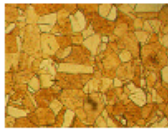

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Learning objects are content items, practice items, and assessment items. These resources are self-contained, digital or non-digital resources that can be used for learning, education or training. Learning objects can be organized to easily form collections of themed content, including traditionally designed courses.

 Laboratory Experiments	 Demonstrations and Techniques	 Exemplars and Case Studies	 Interactive Applications
		 Worksheets	

Un ejemplo de libro abierto a destacar es el: Book: Introduction to Inorganic Chemistry (Wikibook)

Inorganic chemistry is the study of the synthesis, reactions, structures and properties of compounds of the elements. Inorganic chemistry encompasses the compounds - both molecular and extended solids - of everything else in the periodic table, and overlaps with organic chemistry in the area of organometallic chemistry, in which metals are bonded to carbon-containing ligands and molecules. Inorganic chemistry is fundamental to many practical technologies including catalysis and materials (structural, electronic, magnetic etc.), energy conversion and storage, and electronics. Inorganic compounds are also found in biological systems where they are essential to life processes.

 Front Matter	 1: Review of Chemical Bonding	 2: Molecular Orbital Theory	 3: Acid-Base Chemistry
 4: Redox Stability and Redox Reactions	 5: Coordination Chemistry and Crystal Field Theory	 6: Metals and Alloys- Structure, Bonding, Electronic and Magnetic Properties	 7: Metals and Alloys - Mechanical Properties
 8: Ionic and Covalent Solids - Structures	 9: Ionic and Covalent Solids - Energetics	 10: Electronic Properties of Materials - Superconductors and Semiconductors	 11: Basic Science of Nanomaterials
 12: Resources for Students and Teachers	 13: Metals and Alloys - Mechanical Properties	 Back Matter	

Book: Introduction to Inorganic Chemistry (Wikibook) is shared under a CC BY-SA 4.0 license and was authored, remixed, and/or curated by Chemistry 310 via source content that was edited to conform to the style and standards of the LibreTexts platform; a detailed edit history is available upon request.

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Finalmente, hay que señalar que es muy fácil/frecuente acceder a **Chemistry LibreTexts** cuando se realiza una búsqueda en Google de cualquier concepto químico expresado en inglés. Por ejemplo, si buscamos en internet "*splitting of d orbitals in octahedral complexes*" llegaremos a los siguientes resultados:

<https://chem.libretexts.org> › Crystal... ▾ Traducir esta página

Crystal Field Theory - Chemistry LibreTexts



This causes a **splitting** in the energy levels of the **d-orbitals**. This is known as crystal field **splitting**. For **octahedral complexes**, ...

6 may 2021 · Subido por Chuck Wight

[Description of d-Orbitals](#) · [Octahedral Complexes](#) · [Square Planar Complexes](#)

<https://www.toppr.com> › question ▾ Traducir esta página

How do 'd' orbitals split in an octahedral crystal field? - Toppr

In **octahedral** symmetry the **d-orbitals** **split** into two sets with an energy difference, where the $d_{xy}d_{yz}d_{zx}$ orbitals will be lower in energy than the $d_{z^2}d_{x^2-y^2}$...

1 [respuesta](#) · Mejor respuesta: According to Crystal Field Theory, as a ligand approaches the ...